

Designed Synthesis and Assembly of Inorganic Nanomaterials for Energy and Catalytic Applications

Taeghwan Hyeon^{1,2,*}

¹*Institute for Basic Science (IBS), Republic of Korea*

²*Seoul National University, Republic of Korea*

*E-mail address: thyeon@snu.ac.kr

Recently we have focused on the architecture engineering of inorganic nanomaterials for their applications to electrocatalysis and photocatalysis. We present a synthesis of highly durable and active fuel cell electrocatalysts based on ordered M-Pt alloy nanoparticles for oxygen reduction reaction (ORR) in PEMFC. We synthesized highly active and stable electrocatalysts for hydrogen peroxide (H₂O₂) production including Co-N₄(O) moiety incorporated in nitrogen-doped graphene (Co₁-NG(O)) and cobalt phthalocyanine (CoPc) immobilized on oxidized carbon nanotube substrate. The electrocatalytic oxygen evolution reaction (OER) plays a pivotal role in the mass production of hydrogen fuel and chemical feedstocks by various electrochemical reactions. We report that early transition metals with vacant *d* orbitals (*d*⁰-oxoanions) directly participate in and accelerate the alkaline OER via a redox cycle associated with early transition metal-peroxo species. We demonstrate that active machine-learning on even small datasets can discover a “champion” four-metal perovskite oxide OER catalyst. We report on the design and synthesis of highly active TiO₂ photocatalysts incorporated with site-specific single copper atoms that exhibit high photocatalytic hydrogen generation and CO₂ photoreduction activity. We presented a floatable photocatalytic platform constructed from elastomer-hydrogel nanocomposites, demonstrating its superiority over conventional systems in solar hydrogen production. We demonstrated a general method for synthesizing atomically dispersed catalysts via photochemical defect tuning for controlling oxygen vacancy dynamics.

References

- [1] T. Y. Yoo et al., *Energy Environ. Sci.* 16, 1146 (2023).
- [2] M. H. Oh et al., *Nature* 359, 577 (2020).
- [3] J. Jo et al., *Nano Lett.* 22, 3636 (2022).
- [4] J. Kim et al., *Adv. Mater.* 34, 2107868 (2022).
- [5] E. Jung et al., *Nat. Mater.* 19, 436 (2020).
- [6] B. -H. Lee et al., *Nat. Catal.* 6, 234 (2023).
- [7] H. S. Lee et al., *Joule* 7, 1902 (2023).
- [8] K. Lee et al., *Chem* 9, 3600 (2023).
- [9] J. Moon et al., *Nat. Mater.* 23, 108 (2024).
- [10] B. -H. Lee et al., *Energy Environ. Sci.* 15, 601 (2022).
- [11] W. H. Lee et al., *Nat. Nanotechnol.* 18, 754 (2023).
- [12] C. W. Lee et al., *Nat. Mater.* 23, 552 (2024).